



Ultralow-loss and temperature-stable self-composite microwave dielectric ceramic of $\text{Li}_4\text{MgSn}_2\text{O}_7$ – $\text{Li}_2\text{Mg}_3\text{SnO}_6$ for LTCC applications

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ABSTRACT

A self-composite ceramic of $\text{Li}_4\text{MgSn}_2\text{O}_7$ (L_4MS) and $\text{Li}_2\text{Mg}_3\text{SnO}_6$ (L_2MS) phases with compositions $\text{Li}_{2/3(1-x)}\text{Sn}_{1/3(1-x)}\text{Mg}_x\text{O}$ ($x = 1/6$)– y wt.%LiF was for the first time prepared by a conventional solid-state reaction method. The XRD results indicate that L_4MS and $\text{L}_2\text{MS}((\text{MgO})_{\text{ss}})$ phases can stably coexist only as sintering temperature is below 1200 °C, and their relative contents can be regulated by changing sintering temperature. The sintering aid LiF was used to not only reduce sintering temperature to suppress the L_4MS decomposition, but also obtain the dense ceramic body. Excellent microwave dielectric properties of $\epsilon_r = 13.4$, $\text{Qxf} = 123,440$ GHz and near-zero $\tau_f \sim -4.3$ ppm/°C were obtained in the $y = 5$ sample sintered at 1000 °C for 6 h. For LTCC applications, both good chemical compatibility with Ag and optimum microwave dielectric properties of $\epsilon_r = 13.0$, $\text{Qxf} = 78,180$ GHz and $\tau_f = -9$ ppm/°C were achieved in the sample with $y = 6$ as sintered at 850 °C, showing great application potentials in future microwave devices.

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1. Introduction

The demand for low temperature co-fired ceramics (LTCC) materials has gradually increased with the rapid development of LTCC technology for miniaturization and integration of microwave devices. For LTCC applications, microwave dielectric materials should have a low sintering temperature and a super chemical compatibility with Ag electrode, in addition to appropriate dielectric constant ϵ_r , high quality factor Qxf and a near-zero τ_f [1–3]. For this purpose, LTCC materials with excellent dielectric properties have been widely investigated so far [4–9].

Materials with rock-salt structures have attracted extensive attention in recent years for their ultra-low loss. $\text{Li}_2\text{Mg}_3\text{BO}_6$ (L_2MB , $\text{B} = \text{Ti}, \text{Sn}, \text{Zr}$) ceramics possess excellent microwave dielectric properties of $\epsilon_r = 15.2, 8.8$ and 12.6 , $\text{Qxf} = 152,000$ GHz, $123,000$ GHz and $86,000$ GHz, and $\tau_f = -39$ ppm/°C, -32 ppm/°C and -36 ppm/°C, respectively [10]. However, large negative τ_f values have limited their practical applications. A near-zero τ_f value was reported in L_2MB ceramics with the addition of SrTiO_3 or $\text{Ba}_3(\text{VO}_4)_2$ through forming secondary phases with an opposite-sign τ_f value [11,12]. In addition, the porous microstructure

caused by lithium evaporation at higher sintering temperature have significantly deteriorated their microwave dielectric properties. Some efforts have been made to suppress the lithium volatilization in materials containing lithium by means of Li-rich atmosphere protection and adding sintering aids [13–20]. An ultrahigh Qxf value of $\sim 230,000$ – $330,000$ GHz was reported in LiF aided $\text{Li}_2\text{Mg}_3\text{SnO}_6$ (L_2MS) ceramics [15]. Recently, a new pure-phase $\text{Li}_4\text{MgSn}_2\text{O}_7$ (L_4MS) with a positive $\tau_f \sim 12.4$ ppm/°C was successfully synthesized by optimized calcination temperature and time [21], however it can stably exist as sintering temperature is below 1200 °C where unfortunately the sample can be poorly densified. The L_4MS ceramic doped with less than 3 wt% LiF was reported to exhibit $\epsilon_r \sim 13.7$, $\text{Qxf} \sim 97,000$ GHz and $\tau_f \sim -8$ – -13 ppm/°C as sintered at 850 °C.

The phase diagram of the $\text{Li}_{2/3(1-x)}\text{Sn}_{1/3(1-x)}\text{Mg}_x\text{O}$ system was proposed by M. Castellanos and A. R. West in early 1984 [22]. It is indicated from the phase diagram that there is a wide two-phase zone of L_4MS and $\text{L}_2\text{MS}((\text{MgO})_{\text{ss}})$ in the composition range of $x = 1/7$ – $9/10$, in which the relative content of two phases clearly varies with composition and sintering temperature. In this work, a specially designed composition with $x = 1/6$ was investigated for the purpose to achieve a new LTCC ceramic with excellent dielectric properties and particularly a near-zero τ_f by self-compositing L_4MS and L_2MS phases, which means that the synthesized matrix is directly composed of two phases with opposite-sign τ_f directly,

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instead of the addition of a second phase. The phase content was controlled by adjusting appropriate sintering temperature via changing the concentration of sintering aid LiF. The phase evolution, microstructure, microwave dielectric properties of the $x = 1/6$ sample and its chemical compatibility with Ag were studied in detail.

2. Experimental

The $\text{Li}_{2/3(1-x)}\text{Sn}_{1/3(1-x)}\text{Mg}_x\text{O}$ ($x = 1/6$, abbreviated as $\text{LSM}_{1/6}\text{O}$) ceramics were synthesized by a traditional solid-state reaction method using high-purity powders of MgO , SnO_2 and Li_2CO_3 (Sinopharm Chemical Reagent Co. Ltd., Shanghai, China) as starting materials. Stoichiometric amounts of the powders were weighed according to the chemical formula, and were ball-milled for 4 h using zirconia balls and alcohol as the medium on a planetary milling machine. The resulting slurries were then rapidly dried and calcined at 1000°C for 22 h in air. The calcined powders were re-milled for 6 h and then mixed together with 7 wt% PVA as a binder. The granulated powders were subsequently pressed into cylinders with dimensions of 10 mm in diameter and 5–7 mm in height. The specimens were first heated at 550°C in air for 4 h to remove the organic binder, and then sintered at 1100°C – 1250°C for 6 h. Moreover, for LTCC applications and adjustment of the phase content, y wt% LiF sintering aid ($y = 3$ – 10) was added into the as-calcined powder. After compaction, these samples were then sintered in air at 800°C – 1050°C for 6 h at a heating rate of $5^\circ\text{C}/\text{min}$, and at a cooling rate of $10^\circ\text{C}/\text{min}$. To suppress the lithium evaporation, pellets were buried in the powder of the same composition.

The crystal structure of the fired ceramics was identified via an X-ray diffractometer (XRD, D/Max 2500 V, Rigaku, Japan) using $\text{Cu K}\alpha$ radiation. The apparent densities of the sintered ceramics were measured by the Archimedes method. The microstructure of the pellets was observed by using a field-emission scanning electron microscope (FE-SEM; SU8020, JEOL, Tokyo, Japan). For the observation of grain morphology, the samples were polished and then thermally etched at a proper temperature for 15 min. Microwave dielectric properties of the ceramic cylinders were measured using a network analyzer (N5230C, Agilent, Palo Alto, CA) and a temperature chamber (GDW-100, Saiweisi, Changzhou, China). The τ_f values of the samples were measured in the temperature range from 20°C to 80°C and calculated by the following equation:

$$\tau_f = \left(\frac{f_2 - f_1}{T_2 - T_1} \right) f_1$$

where f_1 and f_2 represent the resonant frequencies at T_1 and T_2 , respectively.

3. Results and discussion

Fig. 1 shows the microwave dielectric properties, apparent density and XRD patterns of $\text{LSM}_{1/6}\text{O}$ ceramics sintered at different temperatures for 6 h. It can be seen from Fig. 1(a) that τ_f values tend to be near zero with decreasing sintering temperature. The Qxf values, apparent density and ϵ_r increase gradually with increasing temperature from 1100°C to 1250°C . The variation of microwave dielectric properties with changing sintering temperature can be well understood according to the sample density and the phase evolution. The increase of sample density with sintering temperature will generally help improve ϵ_r and Qxf values. However, L_4MS and $\text{L}_2\text{MS}((\text{MgO})_{\text{ss}})$ phases can stably coexist only at a lower sintering temperature than 1200°C by comparing XRD patterns with standard patterns (JCPDS #37-1164 and JCPDS #39-

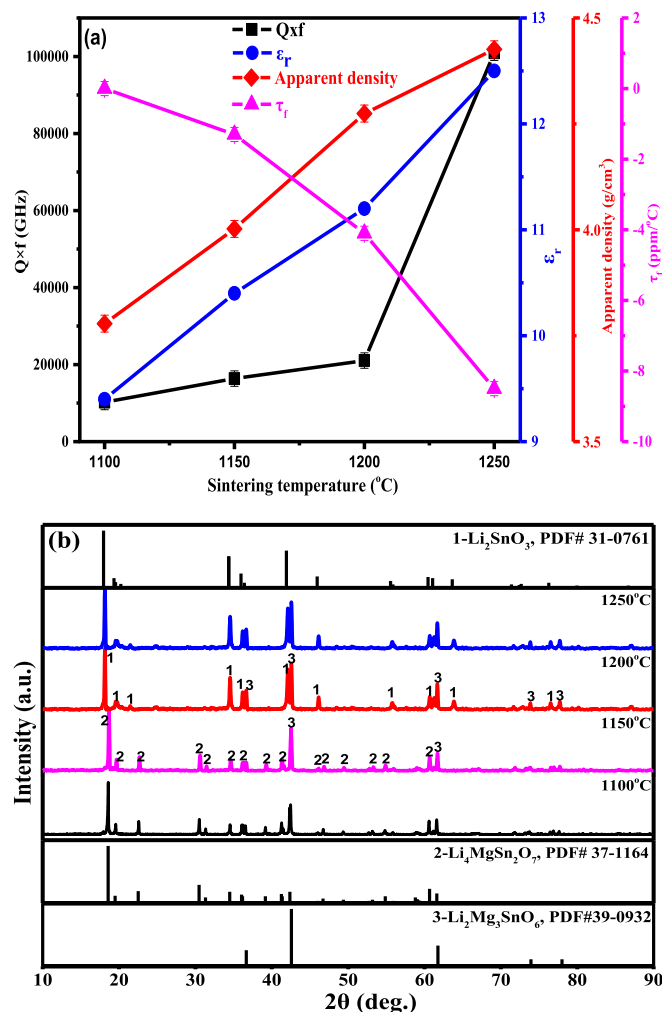


Fig. 1. (a) Microwave dielectric properties and apparent density and (b) XRD patterns of $\text{LSM}_{1/6}\text{O}$ ceramics sintered at different temperatures for 6 h.

0932), as shown in Fig. 1(b). Moreover, the content of $\text{L}_2\text{MS}((\text{MgO})_{\text{ss}})$ phase was found to slightly increase with increasing temperature, keeping consistency with the phase diagram [20], which leads to the shift of the τ_f value towards the side of negative values. There exists a phase transition from L_4MS to $(\text{Li}_2\text{SnO}_3)_{\text{ss}}$ above 1200°C . That is to say, as the sintering temperature is higher 1200°C , the ceramic sample will be composed of $(\text{Li}_2\text{SnO}_3)_{\text{ss}}$ and $\text{L}_2\text{MS}((\text{MgO})_{\text{ss}})$ phases [23]. However, the relative content of $(\text{Li}_2\text{SnO}_3)_{\text{ss}}$ and $\text{L}_2\text{MS}((\text{MgO})_{\text{ss}})$ phases seems insensitive to the rise of sintering temperature, which also fits well to the phase diagram of Li_2SnO_3 and MgO . Considering that pure Li_2SnO_3 ceramics possess microwave dielectric properties of $\epsilon_r = 11.4$, $\tau_f = 14$ ppm/°C and relatively low $\text{Qxf} = 13,100$ GHz [24], it may help deteriorate the Qxf values of the composite ceramic. Owing to large negative values of τ_f for L_2MS , the composite sample sintered at temperatures higher than 1200°C remains a relatively large negative value of τ_f .

For the purpose of reducing sintering temperature and further suppressing the phase transition from L_4MS to Li_2SnO_3 , 3–10 wt% LiF was added into the samples as sintering aids. The apparent density and ϵ_r of $\text{LSM}_{1/6}\text{O}$ -y wt% LiF ceramics sintered at various temperatures are shown in Fig. 2. As sintering temperature increases, the apparent densities of all compositions firstly increase to their respective maximum values approximately at 1000°C

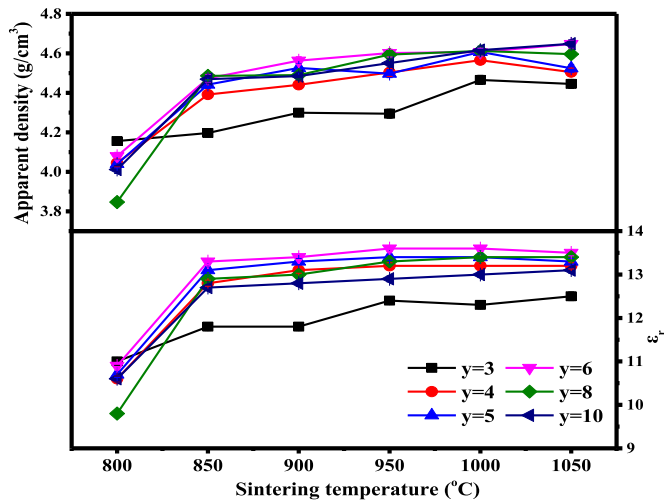


Fig. 2. Apparent density and ϵ_r for $\text{LSM}_{1/6}\text{O}-y$ wt.% LiF ceramics sintered at different temperatures.

and then decreases with further increasing sintering temperature. For the same composition, the variation of ϵ_r with changing sintering temperature presents a tendency similar to that of the apparent density. This result indicates that the sample density should be a dominant factor influencing permittivity. At the same sintering temperature, the difference in permittivity for different compositions should be mainly attributed to the sample density and the content of glass phase owing to the addition of sintering aid LiF. As a result, a relatively low ϵ_r can be achieved approximately at $y = 3$ and $y = 10$. Nevertheless, for samples with $y = 5$ and $y = 6$, the sample density and ϵ_r keep almost constant in the sintering temperature range of 850 °C–1050 °C, indicating that these two samples have been well densified within the above temperature range.

Fig. 3 illustrates the $Q \times f$ value of $\text{LSM}_{1/6}\text{O}-y$ wt.% LiF ceramics sintered at different temperatures. It can be seen that the $Q \times f$ values of each composition firstly increase with increasing sintering temperature, then reach the maximum values, and decrease finally with further increasing temperature. The improvement in $Q \times f$

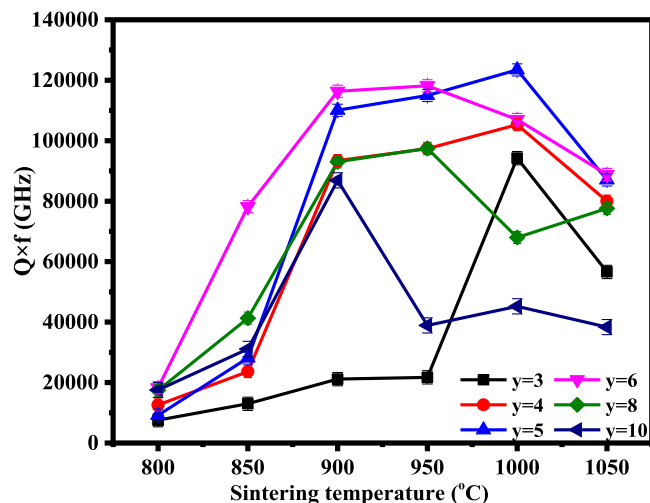


Fig. 3. $Q \times f$ values of $\text{LSM}_{1/6}\text{O}-y$ wt.% LiF ceramics sintered at different temperatures.

values of each composition from 800 °C to 950 °C should be mainly attributed to the increase of sample density. A rapid deterioration of $Q \times f$ values from 1000 °C to 1050 °C may be due to the evaporation of lithium at higher temperatures. The optimal properties shift slightly towards the low temperature as the content of LiF increases. In addition, a higher content of sintering aid LiF might cause the formation of more glass phases in the sample at room temperature after sintering. The SEM images of $\text{LSM}_{1/6}\text{O}-y$ wt.% LiF ($y = 5, 6$) ceramics sintered at 850 °C and 1000 °C are displayed in Fig. 4. It can be seen that these samples have been well densified, indicating LiF doping significantly promotes the sintering behavior. Moreover, the grain size does not distinctly change with sintering temperature. However, as shown in Fig. 4(c), excessive LiF liquid phase (as indicated by arrows) in the $y = 6$ sample can be observed near the grain boundary, as compared with the $y = 5$ sample shown in Fig. 4(b), which would produce negative impacts on $Q \times f$ values. Nevertheless, excellent microwave dielectric properties of $\epsilon_r = 13.4$, $Q \times f = 123,440$ GHz (@ 9.68 GHz) and $\tau_f = -4.3$ ppm/°C were obtained in the $y = 5$ sample sintered at 1000 °C.

The τ_f values of $\text{LSM}_{1/6}\text{O}-y$ wt.% ceramics samples with $y = 5$ and $y = 6$ sintered at different sintering temperatures are shown in Fig. 5. It can be seen that the τ_f values slightly change from -9 ppm/°C to -3 ppm/°C with increasing sintering temperature from 850 °C to 1050 °C probably due to the variation of the phase content. Normalized XRD patterns of $\text{LSM}_{1/6}\text{O}-6$ wt.% LiF ceramics sintered at 800 °C–1050 °C for 6 h are displayed in Fig. 6. All diffraction peaks can be well indexed according to the L_4MS and $\text{L}_2\text{MS}(\text{MgO})_{\text{ss}}$ phases (JCPDS #37-1164 and JCPDS #39-0932) and no any other phases can be observed. The intensity of three major diffraction peaks near 36°, 42° and 62° for the $\text{L}_2\text{MS}(\text{MgO})_{\text{ss}}$ phase apparently increases with increasing sintering temperature from 800 °C to 1050 °C, indicating that the relative content of $\text{L}_2\text{MS}(\text{MgO})_{\text{ss}}$ phase increases. Although the relative content of L_4MS and $\text{L}_2\text{MS}(\text{MgO})_{\text{ss}}$ phases in both samples varies slightly with changing sintering temperature, the $y = 5$ and $y = 6$ samples do not show big differences in τ_f values with changing sintering temperature. This is probably because the τ_f values of $\text{L}_2\text{MS}(\text{MgO})_{\text{ss}}$ and L_4MS phases are close in absolute values, so that the τ_f values of the as-sintered composite ceramic is not very sensitive to their phase contents. For LTCC applications, desirable microwave dielectric properties of $\epsilon_r = 13.0$, $Q \times f = 78,180$ GHz and $\tau_f = -9$ ppm/°C can be achieved in the composite sample with $y = 6$ sintered at 850 °C. In spite of Li-containing compositions in current study, it is noteworthy that they are not sensitive to the humidity since all the samples were placed in air for a few days before the property measurement.

Table 1 presents a comparison of microwave dielectric ceramics for a few Li-based composite ceramics for LTCC applications. It can be seen that the $\text{LSM}_{1/6}\text{O}-6$ wt.% LiF ceramic in current study exhibits larger application potentials because of much higher $Q \times f$ values. To evaluate chemical compatibility with silver, the composite ceramic powder was co-fired with 30 wt% Ag powder at 850 °C for 6 h. As shown in Fig. 7, all diffraction peaks of the co-fired sample can be well indexed to the L_4MS , $\text{L}_2\text{MS}(\text{MgO})_{\text{ss}}$ and Ag, indicating that no any chemical reaction occurs between the composite ceramic and the Ag particle. This could be further confirmed by the SEM image of the co-fired sample, as shown in the inset of Fig. 7. A distinct boundary between the ceramic grains and Ag particles can be observed, indicating silver particles can stably exist in the composite ceramic during sintering. Therefore, the self-composite ceramic of L_4MS and $\text{L}_2\text{MS}(\text{MgO})_{\text{ss}}$ phases reported in this work possesses not only a good chemical compatibility with Ag electrode but also excellent microwave dielectric properties, being expected as a promising candidate material for LTCC applications.

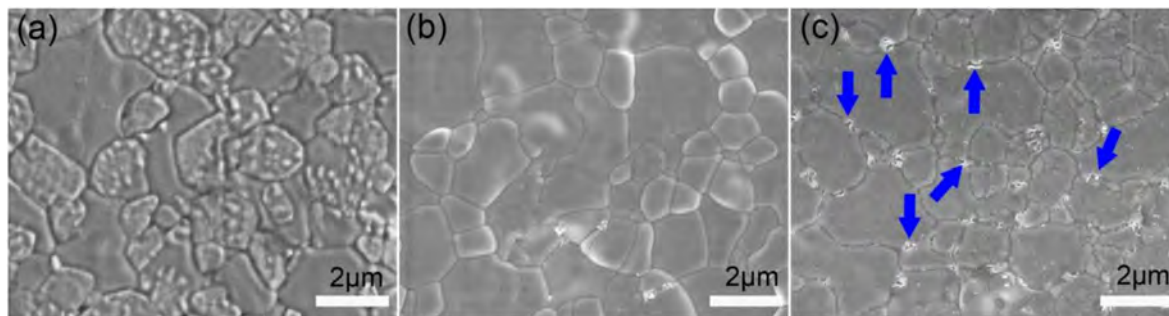


Fig. 4. SEM images of LSM_{1/6}O-y wt.% LiF ceramics sintered at various temperatures: (a) y = 6, at 850 °C (b) y = 5, at 1000 °C (c) y = 6, at 1000 °C.

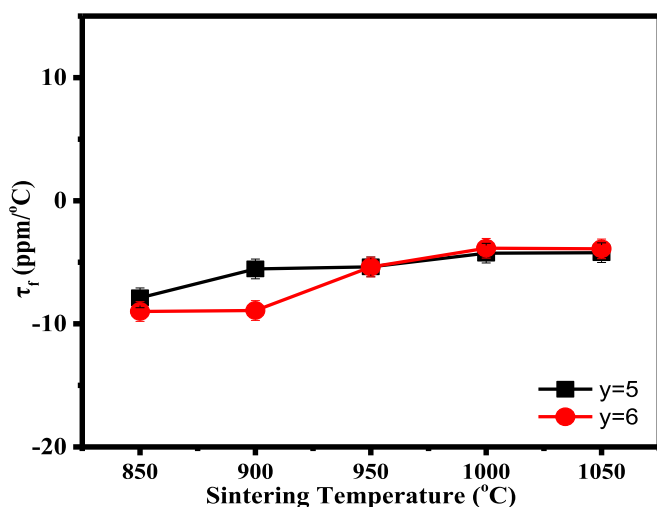


Fig. 5. The τ_r values of LSM_{1/6}O-y wt.% LiF (y = 5-6) ceramics sintered at different temperatures.

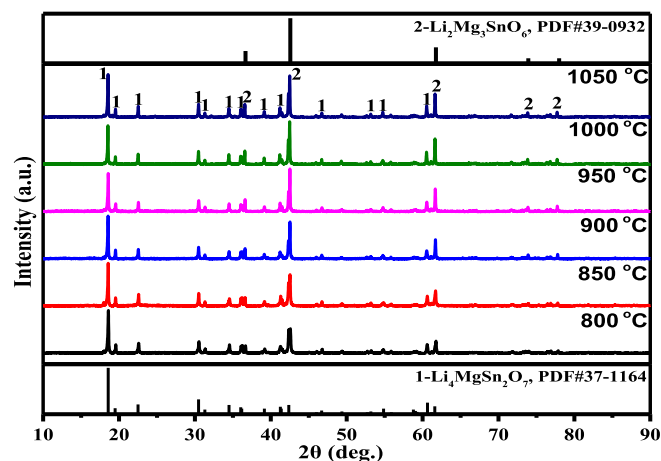


Fig. 6. XRD patterns of LSM_{1/6}O-6 wt.% LiF ceramics sintered at different temperatures for 6 h.

Table 1

Comparison of microwave dielectric properties for a few Li-based composite ceramics for LTCC applications.

Compounds	S.T. (°C)	ϵ_r	Q×f (GHz)	τ_f (ppm/°C)	Ref.
0.86Li ₂ TiO ₃ -0.14Li ₂ CeO ₃	850	21.2	59,310	-7.4	[6]
Li ₂ MgTiO ₄ -10 wt.% LiF	900	18	34,800	4	[19]
-12 wt.% Ca _{0.8} Sr _{0.2} TiO ₃					
Li ₂ ZnTi ₃ O ₈ -4 wt.% TiO ₂	900	26.9	23,560	-1.5	[25]
-1 wt.% CaO-B ₂ O ₃ -SiO ₂					
Li ₂ MgTi ₃ O ₈ -0.5 wt.% BCB	900	26	36,200	-2	[26]
LSM _{1/6} O-6 wt.% LiF	850	13.0	78,180	-9	This work

S.T. Sintering temperature.

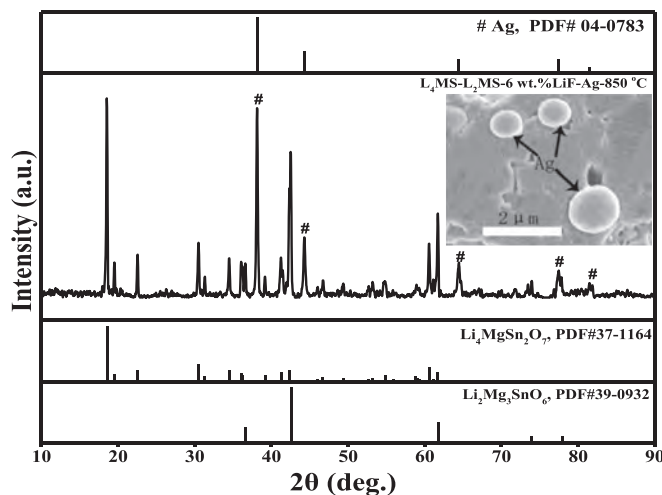


Fig. 7. XRD pattern and SEM image of the LSM_{1/6}O-6 wt.% LiF sample co-fired with Ag at 850 °C for 6 h.

phase as sintered above 1200 °C. Doping LiF sintering aid can not only reduce the sintering temperature to suppress the decomposition of L₄MS phase at high temperature, but also help achieve dense ceramic samples. The relative content of L₂MS((MgO)_{ss}) and L₄MS phases prove to increase with changing sintering temperature, keeping good consistency with the XRD data and the phase diagram. Excellent microwave dielectric properties of $\epsilon_r = 13.4$, $Q \times f = 123,440$ GHz and $\tau_f = -4.3$ ppm/°C were obtained in the sample with y = 5 sintered at 1000 °C. Besides, a high $Q \times f$ value of 78,180 GHz and near-zero τ_f of -9 ppm/°C could be achieved in the sample with y = 6 sintered at 850 °C, together with a good chemical compatibility with Ag, indicating its large potential for LTCC applications.

4. Conclusions

In present work, the phase evolution, sintering behavior and microwave dielectric properties of the LSM_{1/6}O-y wt.% LiF self-composite ceramics have been investigated in detail. The XRD results reveal that L₄MS phase tends to decompose into the Li₂SnO₃

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

CRediT authorship contribution statement

Shuang Zhang: Writing - original draft. **Ruzhong Zuo:** Supervision.

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